

Senior Thesis

**Mixing of Water  
in the Scioto River at  
Chillicothe, Ohio**

by  
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1995

Submitted as partial fulfillment of  
the requirements for the degree of  
Bachelor of Science in Geological Sciences  
at The Ohio State University,  
Spring Quarter, 1995

Approved by:

  
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## ABSTRACT

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A study of Scioto River waters and its tributaries was conducted on August 5, 1994 in and around Chillicothe, Ohio. Samples of main- stream and tributary waters were taken at twelve locations. These sites were chosen to best illustrate mixing of water at confluences, discharge of anthropogenic effluent within city limits, and changes in chemical composition along the main stream. Each sample was analyzed for its concentrations of six conservative elements: Na, Ca, Mg, K, Sr, and S. Element-pair diagrams show mixing of surface water components based on two components. The results indicate that the concentration of water from Paint Creek is  $22.3\% \pm 2.03\%$  of the water in the Scioto River below the confluence. The discharge of Paint Creek was estimated to be  $271 \text{ cfs} \pm 31 \text{ cfs}$  on the day the water samples were collected. Anthropogenic effluent entering Paint Creek acts as a third component of water causing mixtures containing anthropogenic effluent to deviate from the expected mixture of waters of the North and South forks of the river. Data taken along the length of the Scioto River indicate that elemental concentrations vary as a function of distance downstream. Sodium as well as S concentrations increase downstream as a result of anthropogenic inputs into the Scioto River. Ca, Mg, Sr, and K concentrations decrease due to a progressive dilution downstream by addition of dilute water along the course of the river in the study area.

## INTRODUCTION

The study by Hicks (1994) on the chemical composition of water in the Scioto River suggests that the water at Chillicothe merits additional investigation. This current study presents a possible explanation for changes in the chemical composition of the Scioto River at Chillicothe by mixing of water over relatively short distances along the course of the river. In order to determine what is influencing the composition of water in the main stream, two-component mixing models were constructed based on conservative elements plotted against each other. These graphs were interpreted in terms of mixing of Scioto River water and specific tributaries or anthropogenic effluents.

For this purpose, a study of the Scioto River waters and its tributaries was conducted in and around the city of Chillicothe, in central Ross county, Ohio. (See Figure 1) Twelve water samples were collected on August 5, 1994 for chemical analysis. The samples were analyzed for specific conservative elements, these being: K, Na, Ca, Mg, Sr, and total S. Variations of the concentrations of these elements indicate how the water of the Scioto River is affected by the input of water from different sources having different chemical compositions.

The chemical composition of water in the main stream is affected by tributaries discharging into the major stream and

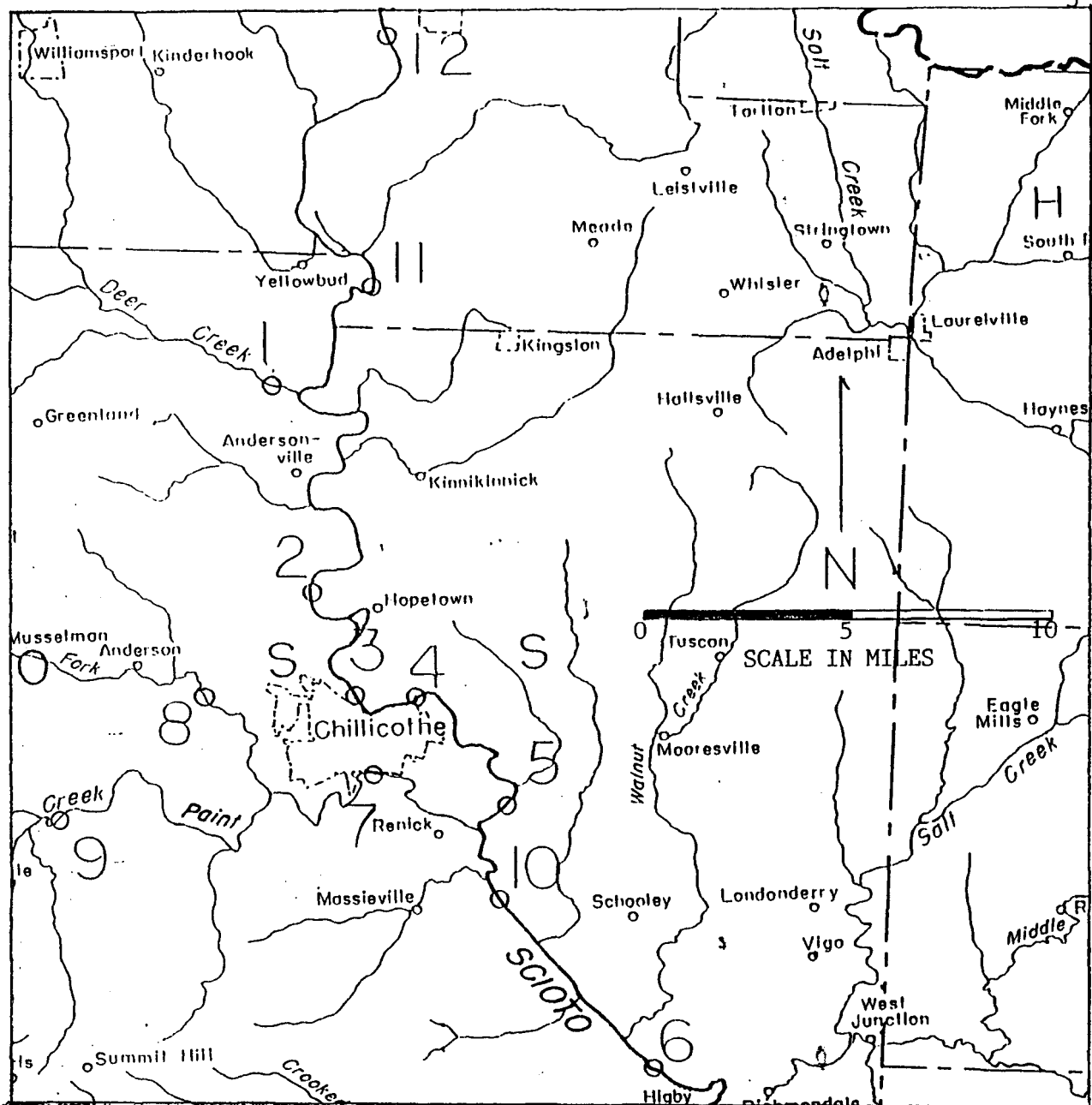


Figure 1. Map of study area around Chillicothe, Ohio.  
Sample locations numbered 1-12.

by anthropogenic effluents discharged at specific locations along the course of the Scioto River and its tributaries.

#### **DESCRIPTION OF COLLECTING LOCATIONS**

Water samples were collected at strategically placed sites to study the compositions of the Scioto River and various tributaries above and below their confluences. In addition, samples were taken along the length of the Scioto River within the study area to investigate the change in the chemical composition of water as a function of distance downstream.

The two major tributaries to the Scioto that were sampled are Paint Creek and Deer Creek. Deer Creek flows into the Scioto River about eight miles north of Chillicothe, whereas Paint Creek enters the Scioto River south of the city. Paint Creek has two branches identified as the Northern and Southern branch, both of which were sampled to test for the presence of anthropogenic sources of water within the city.

#### **FIELD METHODOLOGY AND ANALYTICAL PROCEDURE**

All samples were collected from bridges. At each location, a ten-liter bucket was lowered into the stream from the bridge. A single sample was obtained from each site, taken approximately from the center of the stream. Samples were poured into 500mL Neoprene bottles which were previously rinsed with sample water to prevent contamination. The

samples were then chilled on ice in a cooler until they were transferred to a refrigerator where they remained prior to analysis. In the laboratory, the water was filtered using a 0.45 $\mu$ m Millipore filter under vacuum to remove clay and organic particulates. The samples were then acidified with 1mL of HNO<sub>3</sub> per 250mL aliquot. This preserved the sample by lowering the pH to less than 2. Back-up samples were preserved in the same manner.

Samples were analyzed by Inductively Coupled Plasma Spectrometry and Mass Spectrometry (ICP-MS). In order to ensure the calibration of the ICP-MS, standards for each element were prepared from two stock solutions within the range of concentrations of the unknowns as determined by Hicks (1994). The two stock solutions contained all of the elements being analyzed in this study.

One of the calibration solutions was re-analyzed after every second sample to detect electronic drift in order to assure the accuracy of results. The data indicated that the concentrations of all elements were within the sensitivity limits of the ICP-MS. Selenium, which was also analyzed but not used in this study, was below the sensitivity limits of the ICP-MS.

#### **PRESENTATION OF THE DATA**

The analytical data in Tables 1 to 18 (see appendices) reveal that calcium and sodium are the most abundant elements

in each sample, presumably because of chemical weathering of calcite in carbonate rocks (Ca) and because of subsurface dissolution of Paleozoic salt deposits (Na). The four other elements: Mg, K, S, and Sr occur in substantially lower concentrations than Ca and Na. Magnesium is derived from weathering of dolomite and/or biotite, potassium from both biotite and K-feldspar, sulfur in the form of sulfate is a by-product of the weathering of pyrite and/or dissolution of gypsum, and strontium originates from calcite or celestite. The study area around the city of Chillicothe is underlain by Devonian shale and Mississippian carbonates, shales, and sandstones.

The concentrations of the elements determined in this study are listed in Table 1. Standard deviations are given in Tables 1 to 18 within the appendices.

#### **GRAPHICAL REPRESENTATION OF DATA**

Figures 2 to 4 are graphical representations of the data in the form of two-component mixing diagrams. The numbers correspond to sample sites shown on Figure 1. The concentrations of the elements are indicated along the axes. In each diagram the abscissa is the Na concentrations plotted versus one of the five conservative elements on the ordinate axis in order to preserve continuity among the diagrams. The coordinate axes are scaled to show the greatest detail.

Sample	Element (ppm)					
	Na	Mg	S	K	Ca	Sr
1	7.42	32.1	16.8	1.55	56.7	2.11
2	28.2	17.4	21.4	3.30	48.7	0.92
3	30.6	17.8	22.6	3.53	49.5	0.94
4	28.8	16.6	20.9	3.31	46.0	0.89
5	30.7	18.2	22.6	3.53	49.6	0.93
6	34.2	16.6	23.5	3.06	43.3	0.70
7	61.6	17.7	35.7	3.80	36.7	0.37
8	6.51	26.6	13.0	1.45	48.9	1.46
9	4.32	20.6	6.34	1.67	33.8	0.37
10	36.8	17.5	24.7	3.36	44.5	0.73
11	14.5	24.2	11.4	1.01	57.3	0.57
12	25.7	14.8	18.2	2.89	41.1	0.93

Table 1. Average chemical compositions of water samples from the Scioto River and its tributaries. Sample locations are identified in Figure 1.



Three major confluences were studied around Chillicothe: the Scioto River-Paint Creek confluence, the confluence between the North and South Branches of the Paint Creek, and Scioto River-Deer Creek. The resulting mixing diagrams are presented in figures 2 to 4.

#### **Normal Mixing Model**

The Scioto River-Paint Creek confluence in Figure 2 contains five diagrams containing three data points each. Samples 5 and 10 from the Scioto River represent water from above and below the confluence, respectively. Sample 7 is from the Paint Creek.

The diagrams indicate that each set of samples shows a linear relationship. A line of mixing was constructed by connecting the two sources (5 and 7). Sample 10 represents mixtures of the two components whose concentrations can be determined from its position on the mixing line.

The procedure, illustrated in Figure 2a, involves measuring the distance "a" and dividing it by the distance "b" between the end members. The resulting ratio is converted into a percent representing the concentration of component A in mixture C.

The following data were calculated for the abundance of Paint Creek water (7) in sample 10 (Figure 2) using this method:

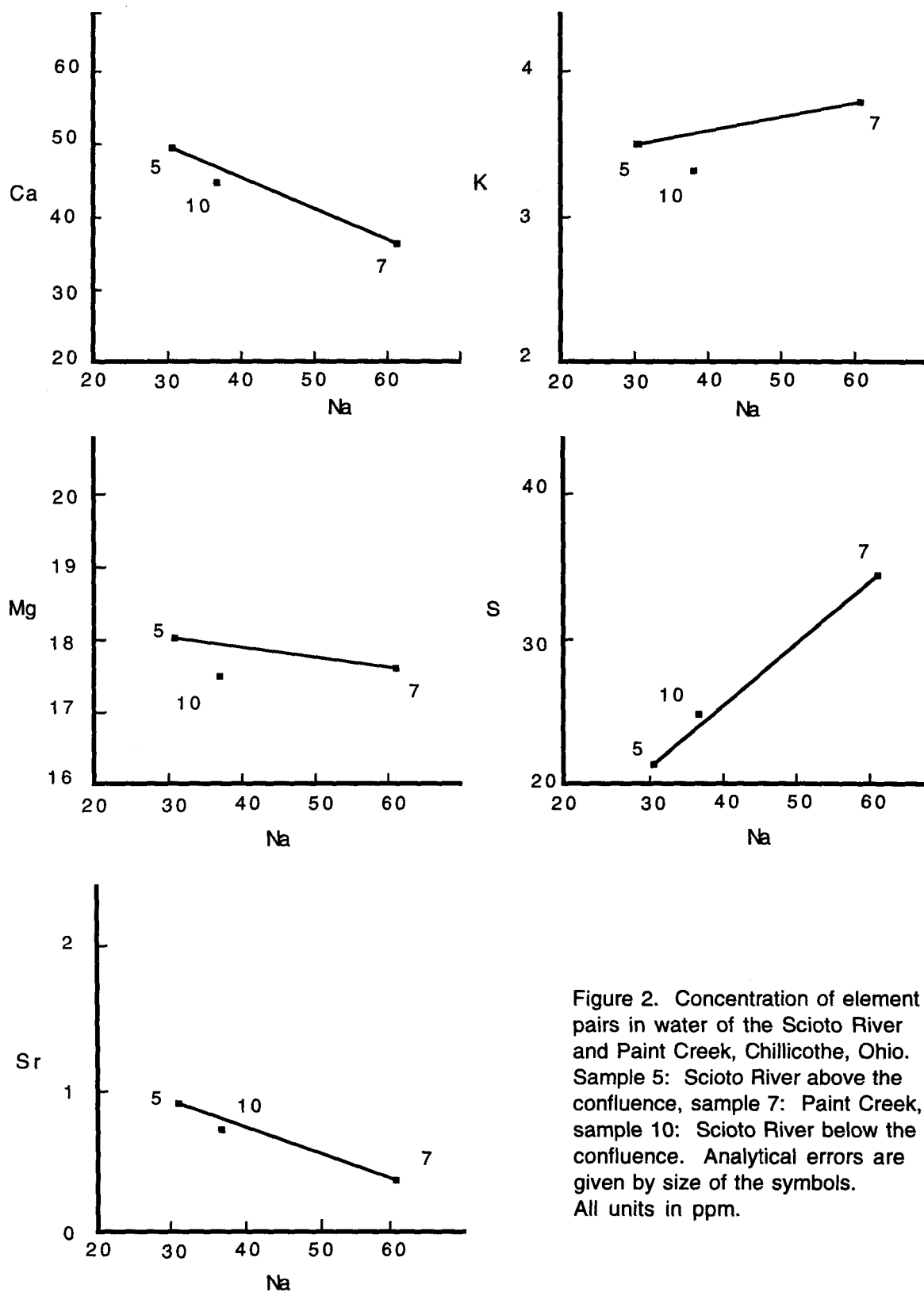


Figure 2. Concentration of element pairs in water of the Scioto River and Paint Creek, Chillicothe, Ohio. Sample 5: Scioto River above the confluence, sample 7: Paint Creek, sample 10: Scioto River below the confluence. Analytical errors are given by size of the symbols. All units in ppm.

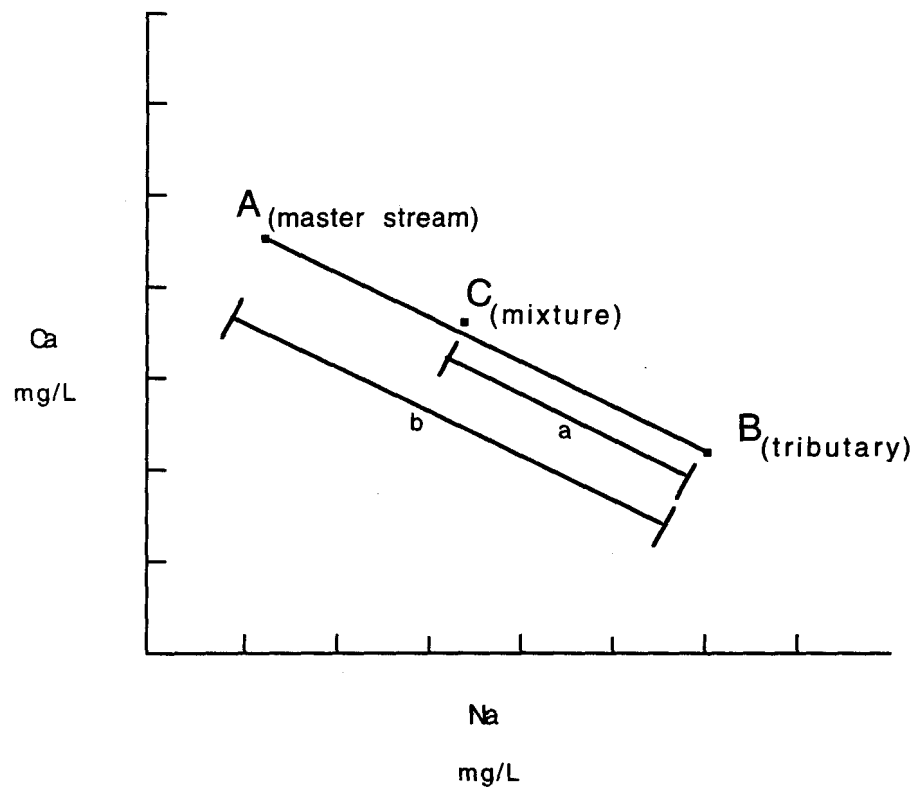


Figure 2a. Sample Na-Ca diagram illustrating the geometric relationship of percent contributions to the mixture from each component. A is the master stream component. B is the tributary component. C is the mixture of the two components. Percent of the tributary component in the mixture is the ratio of  $a/b \times (100)$ .

**Mixing Diagram****Abundance of Paint Creek Water**

Na - Ca	25.0%
Na - Mg	20.0%
Na - S	22.6%
Na - K	23.3%
<u>Na - Sr</u>	<u>20.6%</u>

**Average                      22.3 ± 2.03**

The data indicate that about 22.3% of the water in the Scioto River below the confluence is derived from Paint Creek. The discharge of the Paint Creek can be calculated from this result by knowing the Scioto River discharge above the confluence using the equation:

$$\frac{(\text{Paint Creek discharge})}{(\text{Scioto discharge} + (\text{Paint Creek discharge above confluence}))} = .223 \pm 0.0203 \quad (1)$$

Data obtained from the Water Resources Division of the U.S. Geological Survey reveal that the monthly mean discharge (daily values were not available) of the Scioto River in Chillicothe during August 1994 was 946 cubic feet per second (cfs). This value yields an estimate of 271 cfs ± 31 cfs for the discharge of Paint Creek. The actual mean discharge for August, 1994 was 205 cfs. This means that about 18% of the Scioto waters below the confluence was derived from the Paint Creek. Therefore, the numbers obtained from the mixing model over-estimate the discharge of Paint Creek by an error of 24%.

Samples taken from the northern and southern branches of the Paint Creek (samples 8 and 9), as well as from Paint Creek (sample 7) below the confluence of the two branches are shown in Figure 3. A mixing line was drawn between the two components, but in this case, sample 7, which was taken below the confluence, lies well off the mixing line in the direction of higher Na concentrations. The Na-S diagram shows evidence of increased S concentration as well.

The lack of linearity of the data points in Figure 3 indicates that a third component is affecting sample 7. Since there are no natural inputs of surface water between the confluence and the collecting sites, I conclude that anthropogenic effluent is entering Paint Creek and is altering both the Na and S concentrations of the water.

The Mead paper plant lies within the city limits of Chillicothe and may be the source of the third component of water in Paint Creek. Pulp for the manufacture of paper is prepared by two chemical processes based on the use of either Na sulfate or sulfite to break up cellulose fibers in wood. The discharge of waste water containing these compounds into the Paint Creek can account for increased Na and S concentrations observed in Figure 3.

Information on the chemical composition of the effluent from the Mead plant was obtained from the Ohio EPA, but

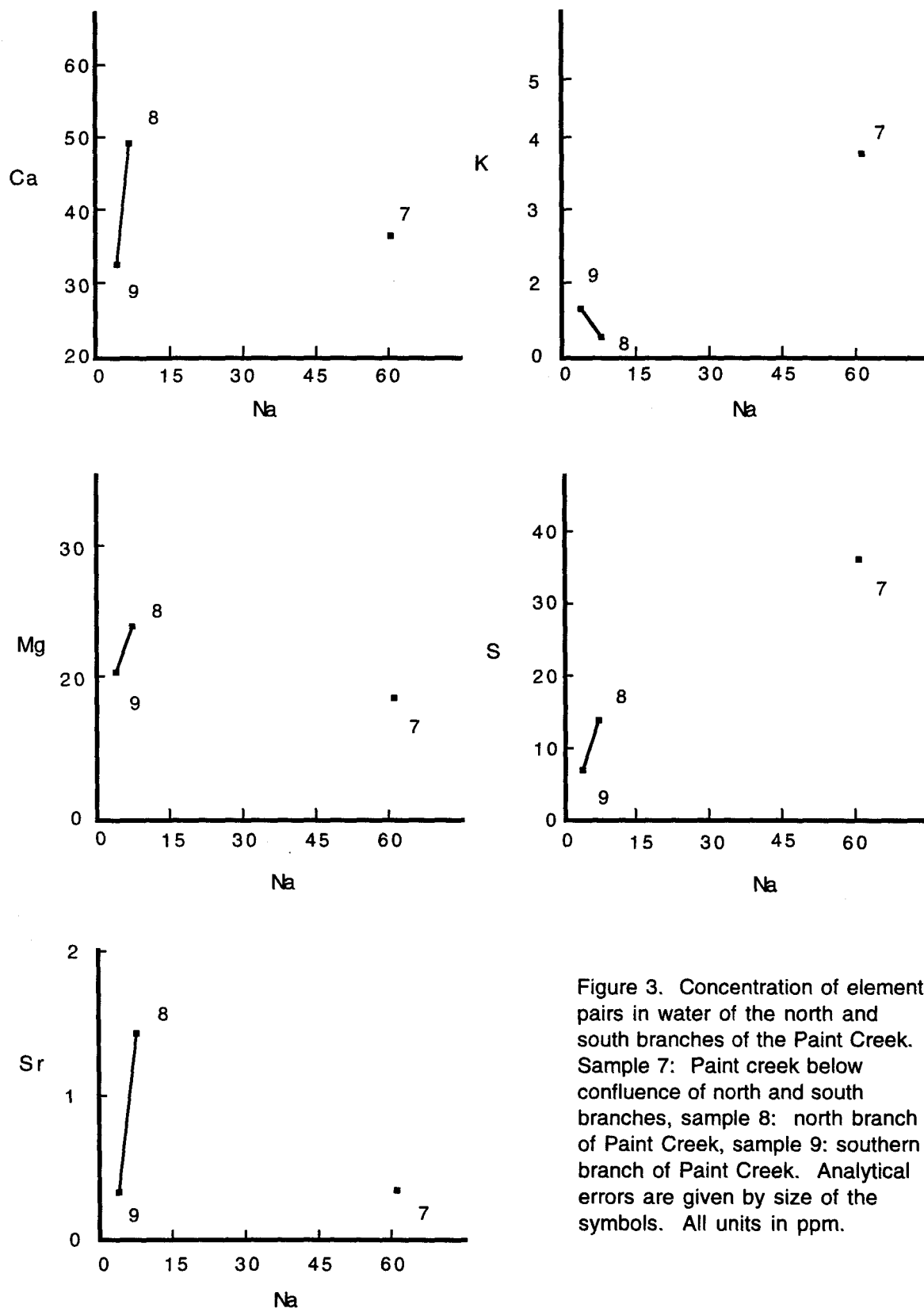


Figure 3. Concentration of element pairs in water of the north and south branches of the Paint Creek. Sample 7: Paint creek below confluence of north and south branches, sample 8: north branch of Paint Creek, sample 9: southern branch of Paint Creek. Analytical errors are given by size of the symbols. All units in ppm.

lacked the Na and S concentrations because these elements are unreactive in solution and are not monitored in effluent as closely as other potentially toxic ions.

### **Scioto River-Deer Creek Confluence**

The Deer Creek-Scioto River diagrams in Figure 4 contain samples 1 (Deer Creek) and 11 (Scioto River above the confluence) as well as samples 2 and 3 below the confluence. These samples (2 and 3) are enriched in Na, S, and K, but depleted in Mg, Ca, and Sr compared to the expected composition predicted by mixing. The diagrams permit the conclusion that a source rich in K, Na, and S has entered the river at a point below the confluence.

The explanation for the deviation of the samples from the predicted mixture is that three tributaries enter the river below the confluence but upstream of the collecting sites on the Scioto River. (see Figure 1) The addition of these waters may increase the concentrations of some elements in the master stream. To study the effects of these tributaries on the composition of water in the Scioto River, more information needs to be obtained from this area.

### **Concentrations as a Function of Distance**

Evidence for the accumulation of Na and S in Scioto River waters is given in Figures 5 and 6. The trend indicates that as water flows downstream, Na and S concentrations rise. Since Na and S are essentially harmless

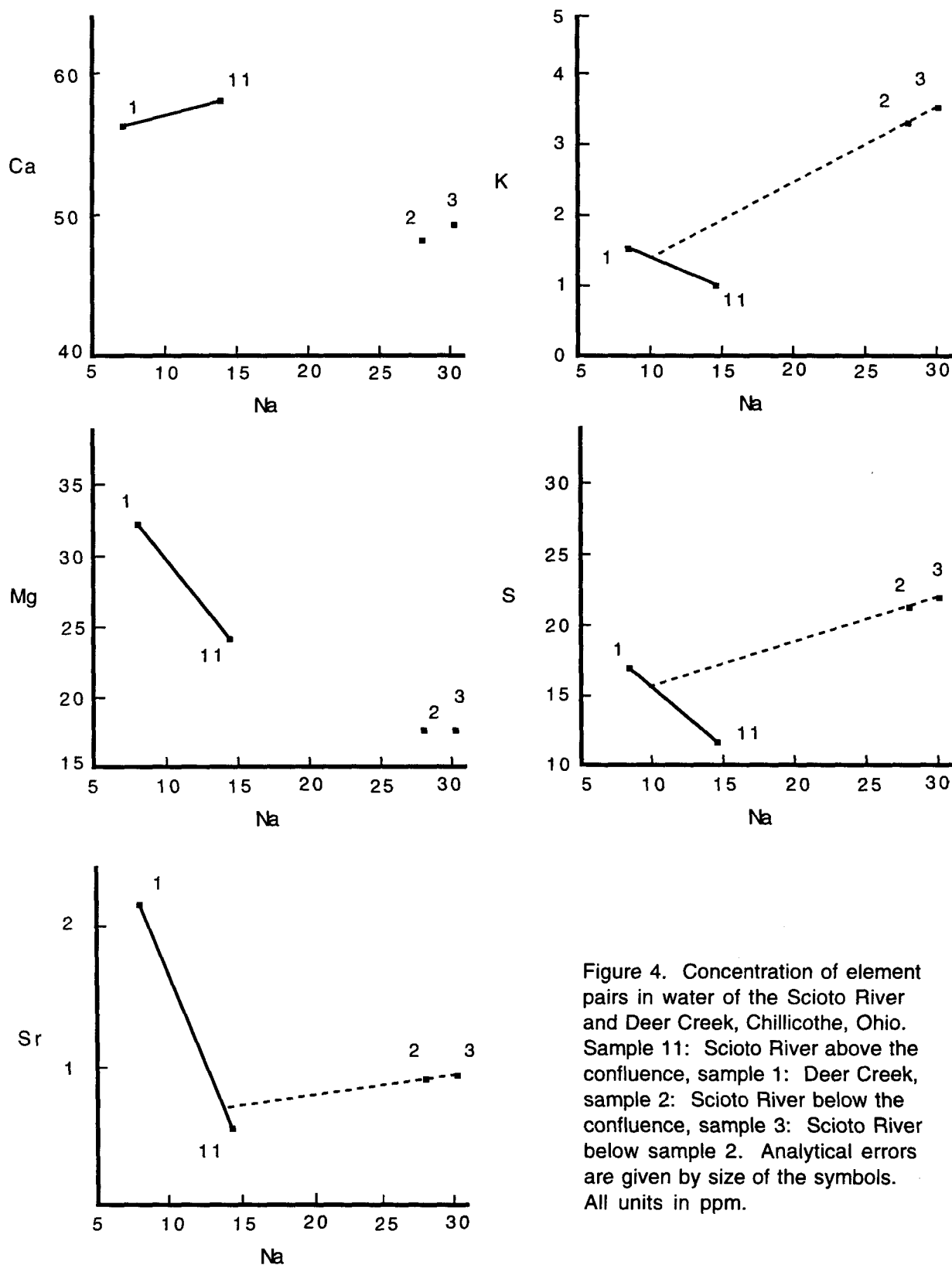


Figure 4. Concentration of element pairs in water of the Scioto River and Deer Creek, Chillicothe, Ohio. Sample 11: Scioto River above the confluence, sample 1: Deer Creek, sample 2: Scioto River below the confluence, sample 3: Scioto River below sample 2. Analytical errors are given by size of the symbols. All units in ppm.



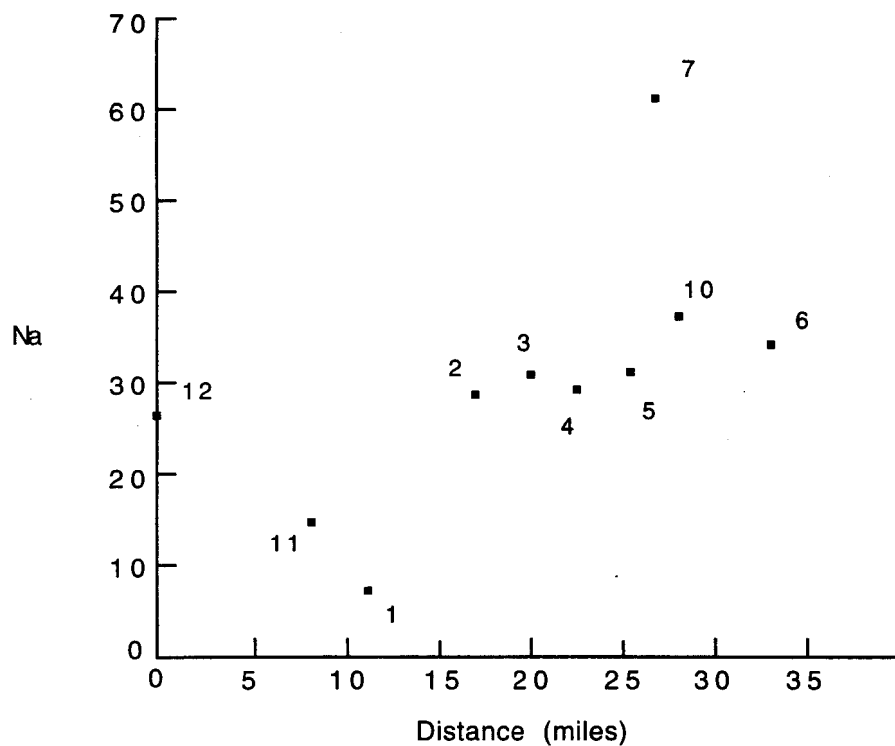


Figure 5. Concentration of sodium in the Scioto River as a function of distance. (Sample 1: Deer Creek concentration, sample 7: Paint Creek concentration). Analytical errors are given by size of the symbol. All concentrations in ppm.

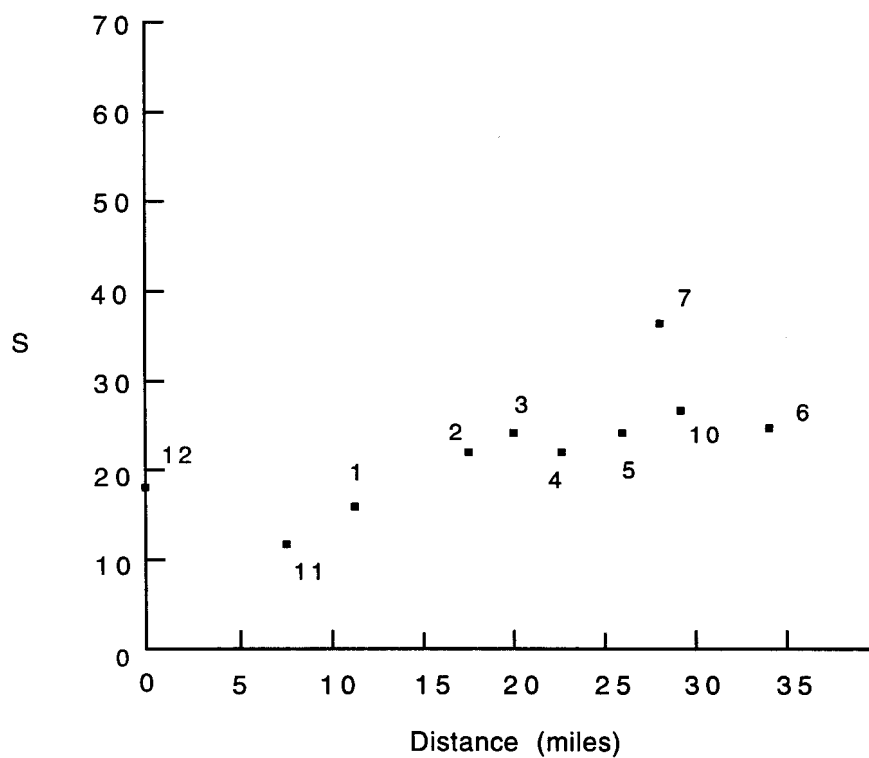


Figure 6. Concentration of total sulfur in the Scioto River as a function of distance. (Sample 1: Deer Creek concentration, sample 7: Paint Creek concentration). Analytical errors are given by size of the symbols. All concentrations in ppm.

to organisms, they are not monitored in industrial effluent and their concentrations therefore pose no threat to public health.

The concentrations of Ca and Mg decrease downstream. (see Figures 7 and 8) I assume that Ca and Mg are not precipitating out, but that their concentrations are being diluted progressively downstream by the increasing discharge of the Scioto River. This hypothesis is confirmed by comparing the discharge of the Scioto River in Chillicothe (946 cfs) and at Higby in southern Ross county (1565 cfs). (USGS, 1994) The increase in discharge of the Scioto River can be attributed to six additional tributaries (including the Paint Creek) that enter the river between Chillicothe and Higby. (see Figure 1)

Therefore, without (anthropogenic) addition of Ca or Mg between Chillicothe and Higby, the concentrations of these elements decrease downstream. The increasing concentrations of Na and S indicate that these elements are overcoming the dilution caused by the increased discharge allowing their concentrations to increase downstream. Therefore, anthropogenic effluent would seem to be the cause of these anomalous trends noted previously by Hicks (1994) who reported increasing Na concentrations for most of the course of the Scioto River.

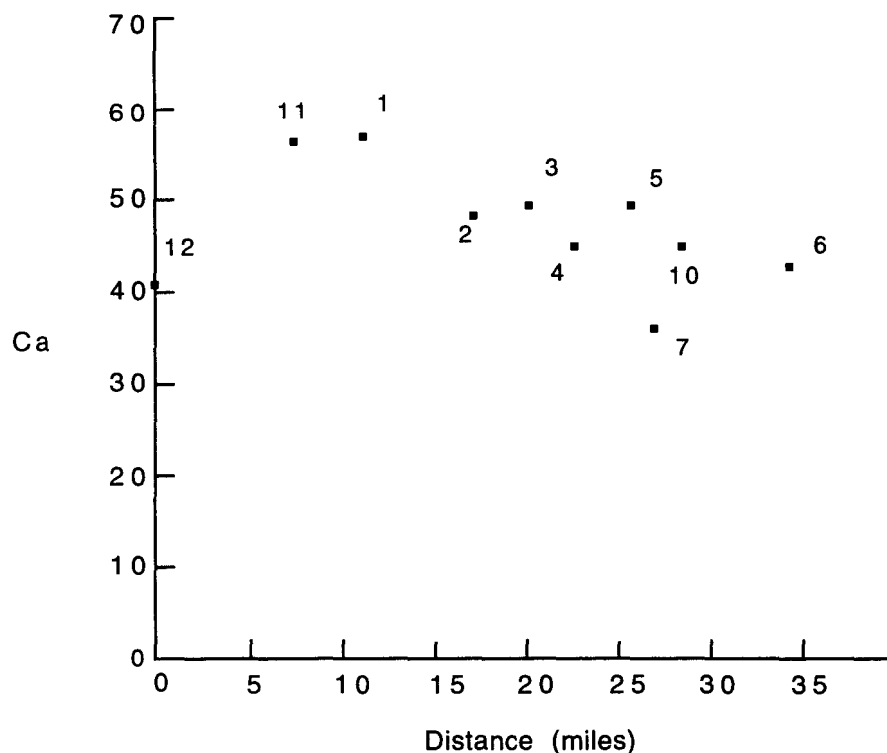


Figure 7. Concentration of calcium in the Scioto River as a function of distance. (Sample 1: Deer Creek concentration, sample 7: Paint Creek concentration). Analytical errors are given by size of the symbol. All concentrations in ppm.

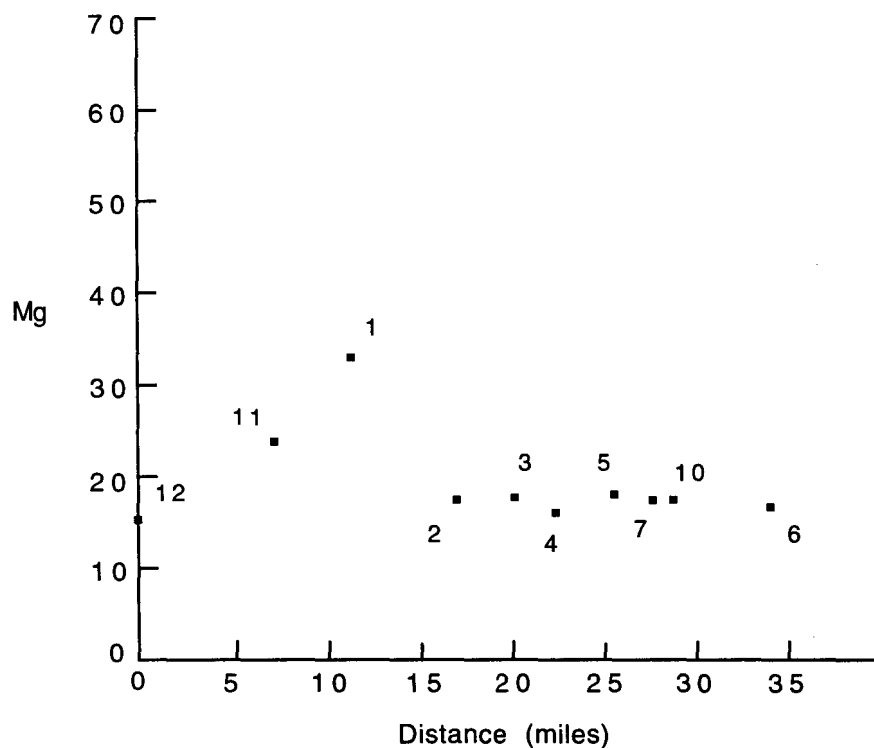


Figure 8. Concentration of magnesium in the Scioto River as a function of distance. (Sample 1: Deer Creek concentration, sample 7: Paint Creek concentration). Analytical errors are given by size of the symbols. All concentrations in ppm.

The chemical composition of water in the Scioto River and its tributaries in and around Chillicothe can be accounted for by mixing of up to three components. I was able to model two-component mixing with data from the Scioto River-Paint Creek confluence. The mixing diagrams were used to estimate the discharge from the Paint Creek into the Scioto River. The resulting estimate is the discharge on the day the samples were collected and therefore differs by about 24% from the average discharge for August, 1994.

Mixing models of the North-South branches of Paint Creek and of the Scioto River-Deer Creek confluence revealed that a third component is affecting the water at these localities. Anthropogenic effluent was found to be contributing Na and S to Paint Creek. The Scioto-Deer Creek confluence requires additional study because the presence of elevated concentrations of Na, K, and S is probably attributable to the discharge of three tributaries below the Deer Creek-Scioto River confluence.

Samples taken along the course of the Scioto River in the study area revealed that concentrations of elements vary with distance downstream: Na and S concentrations increase, whereas Ca and Mg decrease. I propose that dilution is occurring progressively downstream resulting from increased discharge of the Scioto River from Chillicothe to Higby.

(USGS, 1994) Because Ca and Mg are not being added in substantial quantities, either naturally or anthropogenically, their concentrations decline downstream. Sodium and S are added by anthropogenic effluent in amounts that offset, and in fact overcome, the dilution effect.

**ACKNOWLEDGEMENTS**

I thank Dr. Gunter Faure for advising and guiding the research, without whom I would not have been able to initiate or complete such an undertaking. For aiding me in laboratory procedures and analysis, I thank Dr. John Olesik and Kurt Thaxton. Dr. Olesik provided me with the equipment on which to analyze the samples, and Mr. Thaxton helped me with the operation of the ICP-MS. I also thank both Martha Spurbeck and Kathy Jones of the Ohio EPA who made data available to me both graciously and promptly. I thank Ron Velez from the U.S. Geological Survey-Water Resources Division. To my family: Timothy, Sandra, and Trisha; I am ever grateful to you for giving me all of the support I needed during my undergraduate career. Lastly, but certainly not least, I wish to thank Cherie Colton who helped in field research, editing, computer work, and putting up with me for the last year. To her I am forever in debt.

**REFERENCES**

- Hicks, James Edward. Mixing and Anthropogenic Influences on the Chemical Composition of Water Along the Course of the Scioto River, Ohio. Masters Thesis, The Ohio State University, 1994.
- Veley, Ron. Scioto River Basin Discharge in Cubic Feet per Second, October 1993 to September 1994, Daily Values. United States Geological Survey-Water Resources Division, Columbus, Ohio.

**APPENDICES****Output data from the ICP-MS**

(Tables 1 - 18)



**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

24

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 1a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 014  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 14:54:05 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	273965	6616.2622	104699	1020.1544
Mg 24	406262	1.512E+04	146	16.8900
S 34	129713	2187.3203	27015	124.5347
K 39	135352	3857.3369	36208	338.9833
Ca 44	268666	1.028E+04	218	17.1502
Se 78	1804	36.7056	1970	33.2034
Sr 88	657620	2.180E+04	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	169266	6694.4482	7.4285	0.3941 ppm	5.3058
Mg 24	406116	1.512E+04	32.1711	1.3535 ppm	4.2072
S 34	102698	2190.8625	16.8230	0.4282 ppm	2.5451
K 39	99145	3872.2031	1.5575	0.0806 ppm	5.1746
Ca 44	268448	1.028E+04	56.6842	2.1959 ppm	3.8739
Se 78	-166	49.4951	0.0000	n/a ppm	n/a
Sr 88	657550	2.180E+04	2.1122	0.0712 ppm	3.3714

Table 1. Analysis of sample 1.

QUANTITATIVE ANALYSIS: SUMMARY REPORT

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: 2a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 015  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 14:59:33 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	626845	1.170E+04	104699	1020.1544
Mg 24	241066	4193.8706	146	16.8900
S 34	151848	1972.7899	27015	124.5347
K 39	219235	4953.0342	36208	338.9833
Ca 44	231218	5874.3398	218	17.1502
Se 78	1674	24.1356	1970	33.2034
Sr 88	293085	6099.3018	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	522146	1.175E+04	28.2045	0.6917 ppm	2.4526
Mg 24	240921	4193.9048	17.3859	0.3754 ppm	2.1590
S 34	124833	1976.7167	21.1489	0.3863 ppm	1.8266
K 39	183027	4964.6206	3.3034	0.1033 ppm	3.1280
Ca 44	230999	5874.3647	48.6856	1.2547 ppm	2.5771
Se 78	-296	41.0486	0.0000	n/a ppm	n/a
Sr 88	293015	6099.3354	0.9212	0.0199 ppm	2.1630

Table 2. Analysis of sample 2.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

26

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: dilstd  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 016  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:04:16 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample Intensity</b>	<b>Std. Dev.</b>	<b>Blank Intensity</b>	<b>Std. Dev.</b>
Na 23	242407	5025.1499	104699	1020.1544
Mg 24	188611	5804.6440	146	16.8900
S 34	214418	7535.9150	27015	124.5347
K 39	128951	3838.0847	36208	338.9833
Ca 44	148147	6355.8066	218	17.1502
Se 78	6192	168.7850	1970	33.2034
Sr 88	248862	1.259E+04	70	20.3355

	<b>Net Ratio/ Intensity</b>	<b>Std. Dev.</b>	<b>Conc.</b>	<b>Std. Dev.</b>	<b>% RSD</b>
Na 23	137708	5127.6553	5.5705	0.3019 ppm	5.4195
Mg 24	188465	5804.6685	12.6910	0.5195 ppm	4.0936
S 34	187403	7536.9438	33.3771	1.4730 ppm	4.4131
K 39	92744	3853.0254	1.4243	0.0802 ppm	5.6306
Ca 44	147929	6355.8296	30.9428	1.3575 ppm	4.3872
Se 78	4222	172.0199	0.0345	1.709E-03 ppm	4.9543
Sr 88	248792	1.259E+04	0.7768	0.0411 ppm	5.2972

Table 3. Analysis of standard solution for ICP-MSS calibration.

QUANTITATIVE ANALYSIS: SUMMARY REPORT

27

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 3a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 017  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:09:01 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	667376	7958.5093	104699	1020.1544
Mg 24	246075	8205.1309	146	16.8900
S 34	159128	3549.3103	27015	124.5347
K 39	230265	5841.3003	36208	338.9833
Ca 44	234921	8957.4814	218	17.1502
Se 78	1685	27.9971	1970	33.2034
Sr 88	299271	9924.2012	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	562677	8023.6265	30.5908	0.4724 ppm	1.5442
Mg 24	245929	8205.1484	17.8341	0.7344 ppm	4.1178
S 34	132113	3551.4944	22.5717	0.6941 ppm	3.0750
K 39	194057	5851.1279	3.5329	0.1218 ppm	3.4470
Ca 44	234703	8957.4980	49.4765	1.9132 ppm	3.8669
Se 78	-285	43.4315	0.0000	n/a ppm	n/a
Sr 88	299201	9924.2217	0.9415	0.0324 ppm	3.4439

Table 4. Analysis of sample 3.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

28

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: 4a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 018  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:14:03 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	636260	6137.1289	104699	1020.1544
Mg 24	231895	3964.3030	146	16.8900
S 34	150771	702.8731	27015	124.5347
K 39	219582	1889.4757	36208	338.9833
Ca 44	218612	1691.5425	218	17.1502
Se 78	1672	26.7958	1970	33.2034
Sr 88	283508	666.9625	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	531561	6221.3394	28.7589	0.3663 ppm	1.2736
Mg 24	231749	3964.3389	16.5650	0.3548 ppm	2.1419
S 34	123756	713.8204	20.9385	0.1395 ppm	0.6663
K 39	183374	1919.6427	3.3106	0.0400 ppm	1.2069
Ca 44	218394	1691.6294	45.9932	0.3613 ppm	0.7856
Se 78	-298	42.6671	0.0000	n/a ppm	n/a
Sr 88	283438	667.2724	0.8900	2.180E-03 ppm	0.2450

Table 5. Analysis of sample 4.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: dilstd2  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 019  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:18:39 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample Intensity</b>	<b>Std. Dev.</b>	<b>Blank Intensity</b>	<b>Std. Dev.</b>
Na 23	238482	3837.0754	104699	1020.1544
Mg 24	185179	3835.1804	146	16.8900
S 34	210964	5445.7661	27015	124.5347
K 39	126146	3183.1699	36208	338.9833
Ca 44	143834	4805.1904	218	17.1502
Se 78	6197	82.8667	1970	33.2034
Sr 88	241511	8555.1016	70	20.3355

	<b>Net Ratio/ Intensity</b>	<b>Std. Dev.</b>	<b>Conc.</b>	<b>Std. Dev.</b>	<b>% RSD</b>
Na 23	133783	3970.3730	5.3394	0.2338 ppm	4.3780
Mg 24	185033	3835.2175	12.3839	0.3433 ppm	2.7718
S 34	183949	5447.1899	32.7021	1.0645 ppm	3.2553
K 39	89938	3201.1687	1.3659	0.0666 ppm	4.8780
Ca 44	143616	4805.2212	30.0215	1.0263 ppm	3.4187
Se 78	4227	89.2712	0.0346	8.872E-04 ppm	2.5674
Sr 88	241441	8555.1260	0.7528	0.0279 ppm	3.7130

Table 6. Analysis of standard for ICP-MS calibration.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 5a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 020  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:25:17 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	669954	1.691E+04	104699	1020.1544
Mg 24	250044	7441.2031	146	16.8900
S 34	159049	3271.5764	27015	124.5347
K 39	230351	7311.6470	36208	338.9833
Ca 44	235530	8830.0107	218	17.1502
Se 78	1685	12.4301	1970	33.2034
Sr 88	294294	7875.4536	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	565255	1.695E+04	30.7426	0.9977 ppm	3.2452
Mg 24	249898	7441.2222	18.1893	0.6660 ppm	3.6615
S 34	132034	3273.9458	22.5563	0.6398 ppm	2.8366
K 39	194143	7319.5010	3.5347	0.1523 ppm	4.3099
Ca 44	235312	8830.0273	49.6067	1.8860 ppm	3.8019
Se 78	-285	35.4538	0.0000	n/a ppm	n/a
Sr 88	294224	7875.4800	0.9252	0.0257 ppm	2.7810

Table 7. Analysis of sample 5.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

31

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 6a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 021  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:29:59 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	728936	3895.6633	104699	1020.1544
Mg 24	232405	2602.7444	146	16.8900
S 34	163630	651.9040	27015	124.5347
K 39	207310	876.2852	36208	338.9833
Ca 44	205784	1526.8174	218	17.1502
Se 78	1643	26.2151	1970	33.2034
Sr 88	225654	1359.0880	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	624237	4027.0222	34.2152	0.2371 ppm	0.6929
Mg 24	232259	2602.7991	16.6107	0.2330 ppm	1.4024
S 34	136615	663.6925	23.4515	0.1297 ppm	0.5531
K 39	171102	939.5666	3.0552	0.0196 ppm	0.6401
Ca 44	205566	1526.9137	43.2533	0.3261 ppm	0.7540
Se 78	-327	42.3048	0.0000	n/a ppm	n/a
Sr 88	225584	1359.2401	0.7009	4.441E-03 ppm	0.6335

Table 8. Analysis of sample 6.



**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

32

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: dilstd3  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 022  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:35:06 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	235094	4311.1138	104699	1020.1544
Mg 24	180444	5002.6768	146	16.8900
S 34	206260	6171.6475	27015	124.5347
K 39	123724	3317.2346	36208	338.9833
Ca 44	140346	5426.7373	218	17.1502
Se 78	6083	83.8836	1970	33.2034
Sr 88	233797	9040.2324	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	130395	4430.1714	5.1400	0.2608 ppm	5.0745
Mg 24	180298	5002.7051	11.9601	0.4477 ppm	3.7437
S 34	179245	6172.9038	31.7828	1.2064 ppm	3.7957
K 39	87516	3334.5098	1.3155	0.0694 ppm	5.2759
Ca 44	140128	5426.7646	29.2765	1.1591 ppm	3.9591
Se 78	4112	90.2159	0.0334	8.965E-04 ppm	2.6828
Sr 88	233727	9040.2549	0.7275	0.0295 ppm	4.0595

Table 9. Analysis of standard solution for ICP-MS calibration.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

33

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 7a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 023  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:40:01 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample Intensity</b>	<b>Std. Dev.</b>	<b>Blank Intensity</b>	<b>Std. Dev.</b>
Na 23	1194375	2.370E+04	104699	1020.1544
Mg 24	244461	6293.7100	146	16.8900
S 34	226386	4515.8574	27015	124.5347
K 39	242950	6493.2251	36208	338.9833
Ca 44	174974	4277.3818	218	17.1502
Se 78	1668	27.2791	1970	33.2034
Sr 88	123959	2788.2439	70	20.3355

	<b>Net Ratio/ Intensity</b>	<b>Std. Dev.</b>	<b>Conc.</b>	<b>Std. Dev.</b>	<b>% RSD</b>
Na 23	1089676	2.373E+04	61.6182	1.3968 ppm	2.2669
Mg 24	244315	6293.7324	17.6897	0.5633 ppm	3.1843
S 34	199371	4517.5742	35.7161	0.8829 ppm	2.4719
K 39	206742	6502.0674	3.7970	0.1353 ppm	3.5642
Ca 44	174756	4277.4160	36.6727	0.9136 ppm	2.4912
Se 78	-302	42.9722	0.0000	n/a ppm	n/a
Sr 88	123889	2788.3181	0.3687	9.110E-03 ppm	2.4707

Table 10. Analysis of sample 7.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 8a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 024  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:44:35 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample Intensity</b>	<b>Std. Dev.</b>	<b>Blank Intensity</b>	<b>Std. Dev.</b>
Na 23	258379	1959.7562	104699	1020.1544
Mg 24	343713	3796.6919	146	16.8900
S 34	110238	1598.9337	27015	124.5347
K 39	130344	1962.7412	36208	338.9833
Ca 44	232196	4873.5132	218	17.1502
Se 78	1645	14.8024	1970	33.2034
Sr 88	457786	1.098E+04	70	20.3355

	<b>Net Ratio/ Intensity</b>	<b>Std. Dev.</b>	<b>Conc.</b>	<b>Std. Dev.</b>	<b>% RSD</b>
Na 23	153680	2209.3799	6.5109	0.1301 ppm	1.9979
Mg 24	343567	3796.7295	26.5728	0.3398 ppm	1.2788
S 34	83223	1603.7761	13.0170	0.3134 ppm	2.4078
K 39	94137	1991.7988	1.4532	0.0415 ppm	2.8526
Ca 44	231978	4873.5435	48.8947	1.0409 ppm	2.1289
Se 78	-325	36.3534	0.0000	n/a ppm	n/a
Sr 88	457716	1.098E+04	1.4593	0.0359 ppm	2.4586

Table 11. Analysis of sample 8.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: dilstd4  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 025  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:49:46 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample</b>	<b>Std.</b>	<b>Blank</b>	<b>Std.</b>
	<b>Intensity</b>	<b>Dev.</b>	<b>Intensity</b>	<b>Dev.</b>
Na 23	247531	2717.1157	104699	1020.1544
Mg 24	192716	3524.2930	146	16.8900
S 34	223420	3890.6143	27015	124.5347
K 39	132943	2163.0303	36208	338.9833
Ca 44	156038	3534.9878	218	17.1502
Se 78	6480	119.4783	1970	33.2034
Sr 88	260947	8687.6650	70	20.3355

	<b>Net Ratio/</b>	<b>Std.</b>		<b>Std.</b>	
	<b>Intensity</b>	<b>Dev.</b>	<b>Conc.</b>	<b>Dev.</b>	<b>% RSD</b>
Na 23	142832	2902.3152	5.8722	0.1709 ppm	2.9099
Mg 24	192570	3524.3335	13.0584	0.3154 ppm	2.4155
S 34	196405	3892.6069	35.1364	0.7607 ppm	2.1651
K 39	96735	2189.4314	1.5073	0.0456 ppm	3.0232
Ca 44	155820	3535.0293	32.6283	0.7550 ppm	2.3141
Se 78	4510	124.0061	0.0374	1.232E-03 ppm	3.2980
Sr 88	260877	8687.6885	0.8162	0.0284 ppm	3.4773

Table 12. Analysis of standard solution for ICP-MS calibration.

QUANTITATIVE ANALYSIS: SUMMARY REPORT

36

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: ~~8a~~ 9a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 026  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 15:59:27 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	221126	3236.2219	104699	1020.1544
Mg 24	277026	5360.5356	146	16.8900
S 34	76065	1027.8345	27015	124.5347
K 39	140670	2531.0117	36208	338.9833
Ca 44	161406	4103.2427	218	17.1502
Se 78	1715	36.7185	1970	33.2034
Sr 88	125176	3024.7854	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	116427	3393.2061	4.3176	0.1998 ppm	4.6271
Mg 24	276880	5360.5620	20.6043	0.4798 ppm	2.3285
S 34	49050	1035.3514	6.3386	0.2023 ppm	3.1922
K 39	104462	2553.6111	1.6682	0.0531 ppm	3.1861
Ca 44	161188	4103.2783	33.7748	0.8764 ppm	2.5949
Se 78	-256	49.5047	0.0000	n/a ppm	n/a
Sr 88	125106	3024.8538	0.3727	9.882E-03 ppm	2.6517

Table 13. Analysis of sample 9.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

37

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118

Sample ID: 10a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 027  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 16:04:59 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35

Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample Intensity</b>	<b>Std. Dev.</b>	<b>Blank Intensity</b>	<b>Std. Dev.</b>
Na 23	772827	7193.7407	104699	1020.1544
Mg 24	242745	2194.1338	146	16.8900
S 34	169795	638.6733	27015	124.5347
K 39	222134	1995.2152	36208	338.9833
Ca 44	211500	3309.8889	218	17.1502
Se 78	1668	41.7778	1970	33.2034
Sr 88	233907	2298.2847	70	20.3355

	<b>Net Ratio/ Intensity</b>	<b>Std. Dev.</b>	<b>Conc.</b>	<b>Std. Dev.</b>	<b>% RSD</b>
Na 23	668128	7265.7153	36.7993	0.4278 ppm	1.1625
Mg 24	242599	2194.1987	17.5361	0.1964 ppm	1.1199
S 34	142780	650.7016	24.6564	0.1272 ppm	0.5158
K 39	185927	2023.8066	3.3637	0.0421 ppm	1.2523
Ca 44	211282	3309.9333	44.4741	0.7070 ppm	1.5896
Se 78	-302	53.3652	0.0000	n/a ppm	n/a
Sr 88	233837	2298.3745	0.7279	7.509E-03 ppm	1.0316

Table 14. Analysis of sample 10.

QUANTITATIVE ANALYSIS: SUMMARY REPORT

38

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: dilstd5  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 028  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 16:09:53 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	237160	4770.8779	104699	1020.1544
Mg 24	186308	4210.2275	146	16.8900
S 34	210724	6169.7559	27015	124.5347
K 39	124873	3407.0149	36208	338.9833
Ca 44	143496	5609.7754	218	17.1502
Se 78	6169	89.5157	1970	33.2034
Sr 88	238654	1.108E+04	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	132461	4878.7285	5.2616	0.2872 ppm	5.4592
Mg 24	186162	4210.2612	12.4849	0.3768 ppm	3.0182
S 34	183709	6171.0127	32.6552	1.2060 ppm	3.6932
K 39	88665	3423.8372	1.3394	0.0713 ppm	5.3205
Ca 44	143278	5609.8018	29.9494	1.1982 ppm	4.0007
Se 78	4199	95.4753	0.0343	9.488E-04 ppm	2.7680
Sr 88	238584	1.108E+04	0.7434	0.0362 ppm	4.8681

Table 15. Analysis of standard solution for ICP-MS calibration.

QUANTITATIVE ANALYSIS: SUMMARY REPORT

39

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: 11a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 029  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 16:15:16 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	394319	7716.3823	104699	1020.1544
Mg 24	316697	7558.1367	146	16.8900
S 34	101899	1979.5649	27015	124.5347
K 39	109010	2185.4353	36208	338.9833
Ca 44	271318	8030.7314	218	17.1502
Se 78	1659	17.2259	1970	33.2034
Sr 88	185513	4530.6143	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	289620	7783.5254	14.5144	0.4583 ppm	3.1573
Mg 24	316551	7558.1558	24.1549	0.6765 ppm	2.8005
S 34	74884	1983.4783	11.3875	0.3876 ppm	3.4040
K 39	72802	2211.5688	1.0092	0.0460 ppm	4.5610
Ca 44	271099	8030.7500	57.2504	1.7153 ppm	2.9961
Se 78	-311	37.4058	0.0000	n/a ppm	n/a
Sr 88	185443	4530.6597	0.5698	0.0148 ppm	2.5977

Table 16. Analysis of sample 11.



QUANTITATIVE ANALYSIS: SUMMARY REPORT

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: 12a  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 030  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 16:20:05 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	Sample Intensity	Std. Dev.	Blank Intensity	Std. Dev.
Na 23	584384	2438.6206	104699	1020.1544
Mg 24	212444	2706.7671	146	16.8900
S 34	136542	992.7105	27015	124.5347
K 39	199332	1493.5933	36208	338.9833
Ca 44	195663	1366.2294	218	17.1502
Se 78	1702	28.4179	1970	33.2034
Sr 88	296764	2028.9023	70	20.3355

	Net Ratio/ Intensity	Std. Dev.	Conc.	Std. Dev.	% RSD
Na 23	479685	2643.4041	25.7046	0.1556 ppm	0.6055
Mg 24	212298	2706.8198	14.8241	0.2423 ppm	1.6342
S 34	109527	1000.4914	18.1576	0.1955 ppm	1.0768
K 39	163125	1531.5778	2.8891	0.0319 ppm	1.1034
Ca 44	195445	1366.3370	41.0917	0.2918 ppm	0.7102
Se 78	-268	43.7040	0.0000	n/a ppm	n/a
Sr 88	296694	2029.0043	0.9333	6.629E-03 ppm	0.7103

Table 17. Analysis of sample 12.

**QUANTITATIVE ANALYSIS: SUMMARY REPORT**

41

Data Set: kkt-geoH2O1118  
 Data Set Description: Analysis of Scioto & trib. water; undergrad thesis  
 Parameter File: kkt-geoH2O1118  
  
 Sample ID: dilstd6  
 Sample Description:  
 Sample Type: Sample  
 Sequence Number: 031  
 Blank: Subtracted (010)  
 Dilution Factor: 1  
 Number of Repeats: 6  
 Time: 16:25:13 Nov 18 1994  
 Signal Profile Processing: Average  
 Spectral Peak Processing: Average  
 Deadtime Correction: 35  
  
 Calibration File: kkt-geoH2O-1  
 Calibration: External Standard

	<b>Sample</b>	<b>Std.</b>	<b>Blank</b>	<b>Std.</b>
	<b>Intensity</b>	<b>Dev.</b>	<b>Intensity</b>	<b>Dev.</b>
Na 23	233936	3818.6008	104699	1020.1544
Mg 24	184680	5092.9907	146	16.8900
S 34	202898	6992.0889	27015	124.5347
K 39	121230	3398.9167	36208	338.9833
Ca 44	137528	5737.7085	218	17.1502
Se 78	6228	107.6924	1970	33.2034
Sr 88	227939	9238.8730	70	20.3355

	<b>Net Ratio/</b>	<b>Std.</b>		<b>Std.</b>	
	<b>Intensity</b>	<b>Dev.</b>	<b>Conc.</b>	<b>Dev.</b>	<b>% RSD</b>
Na 23	129237	3952.5217	5.0718	0.2327 ppm	4.5883
Mg 24	184534	5093.0186	12.3392	0.4558 ppm	3.6942
S 34	175883	6993.1978	31.1258	1.3667 ppm	4.3909
K 39	85022	3415.7788	1.2636	0.0711 ppm	5.6265
Ca 44	137310	5737.7339	28.6747	1.2255 ppm	4.2738
Se 78	4258	112.6948	0.0349	1.120E-03 ppm	3.2120
Sr 88	227869	9238.8955	0.7084	0.0302 ppm	4.2608

Table 18. Analysis of standard solution for ICP-MS calibration.